

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance  
 (B) volume, molarity, pH, amount of substance  
 (C) volume, molarity, pH, concentration of ions  
 (D) volume, molarity, pH, concentration of ions, amount of substance

(A)  $V = \text{dm}^3$ ,  $M = \text{mol dm}^{-3}$

(B)  $p = 10^{-\text{pH}}$ ,  $V = \text{dm}^3$ ,  $M = \text{mol dm}^{-3}$ ,  $n = \text{mol}$   
 (C)  $V = \text{dm}^3$ ,  $p = 10^{-\text{pH}}$ ,  $M = \text{mol dm}^{-3}$ ,  $n = \text{mol}$  (not all the same)

2. An acid-base indicator has a  $\text{p}K_{\text{a}}$  of 4.5. The indicator is colourless in its protonated form and pink in its deprotonated form. What colour will the indicator be in a solution of pH 4.5? (correct)

(A)  $\text{pH} < \text{p}K_{\text{a}}$ , colourless  
 (B)  $\text{pH} > \text{p}K_{\text{a}}$ , pink

(C)  $\text{pH} = \text{p}K_{\text{a}}$ , pink  
 (D)  $\text{pH} = \text{p}K_{\text{a}}$ , colourless  
 (E)  $\text{pH} = \text{p}K_{\text{a}}$ , pink

3. An acid-base indicator has a  $\text{p}K_{\text{a}}$  of 4.5. The indicator is colourless in its protonated form and pink in its deprotonated form. What colour will the indicator be in a solution of pH 5.5? (correct)

(A) pink  
 (B) colourless  
 (C) pink  
 (D) colourless

4. Which indicator is most appropriate for the following reaction? (correct)



(A) pH 10  
 (B) pH 7  
 (C) pH 4  
 (D) pH 2

5. Amino acids contain both an acidic group and a basic group in their structure. In a pH of approximately 7.4, the amino acid is zwitterionic. What does zwitterionic mean? (correct)



(A)  $\text{NH}_3^+$  and  $\text{COO}^-$   
 (B)  $\text{NH}_2$  and  $\text{COOH}$   
 (C)  $\text{NH}_3^+$  and  $\text{COOH}$   
 (D)  $\text{NH}_2$  and  $\text{COO}^-$   
 (E)  $\text{NH}_3^+$  and  $\text{NH}_2$

6. An acid-base equilibrium,  $\text{CH}_3\text{COOH} + \text{H}_2\text{O} \rightleftharpoons \text{CH}_3\text{COO}^- + \text{H}_3\text{O}^+$ , is shown below. Which of the following is the correct expression for the acid dissociation constant,  $K_{\text{a}}$ ? (correct)



# Multiple Choice Stoichiometry Test

**CL Gary**



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