

**MID TERM 2 2023 EXAM****CHEMISTRY FORM 2****TIME: 1 3/4 HOURS****NAME \_\_\_\_\_ ADM \_\_\_\_\_ CLASS \_\_\_\_\_**

1. The table below gives information on four elements by letters K, L, M and N. Study it and answer the questions that follow. The letters do not represent the actual symbols of the elements.

ELEMENT	Electron arrangement	Atomic radius nm	Ionic radius nm	
K	2.8.2	0.036	0.065	
L	2.8.7	0.099	0.181	
M	2.8.8.1	0.203	0.133	
N	2.8.8.2	0.174	0.099	

a) Which two elements have two similar properties? Explain. (2mks)

.....

.....

b) What is most likely formula of oxide of L? (1mk)

.....

.....

c) Which element is a non-metal? Explain (2mks)

.....

.....

d) Which one of elements is the strongest reducing agent? Explain (2mks)

.....

.....

e) Explain why ionic radius of N is less than that of M (2mks)

.....

.....

f) Explain why the ionic radius of L is bigger than its atomic radius (1mk)

.....

.....

2. a) Define the following terms as used in chemistry (5mks)

i. Salt

.....

ii. Hygroscopy

.....

iii. Deliquescent salts

# Nd Semester Exam Chemistry Form

**Arturo Cuomo**



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