

THE SKELETAL SYSTEM

TASK 1. MULTIPLE CHOICE

- The bones of the skeletal system can be subdivided into two large groups, the _____ skeleton, and the _____ skeleton.
A. pelvic / appendicular
B. axial/ appendicular
C. axial/ pectoral
D. pectoral / pelvic
- The skull is subdivided into two parts: the _____ and the _____ portion.
A. cranium / facial
B. cranial / face
C. cranial/ facial
D. cranium/ face
- The _____ are the two bones that form the upper jaw.
A. mandibles
B. maxillae
C. mandible
D. maxilla
- The _____ are the two bones that form the lower jaw.
A. mandibles
B. maxillae
C. mandible
D. maxilla
- The zygomatic bones are also called the _____.
A. jawbones
B. wrist bones
C. thigh bones
D. cheekbones
- Skull bones are mostly connected by _____.
A. ligaments
B. tendons
C. sutures
D. cartilages
- Four or five small terminal vertebrae are also fused together to form the _____.
A. coccyx
B. sacrum
C. atlas
D. axis
- The _____ is composed of twelve pairs of ribs and the _____.
A. sternum/ thorax
B. thorax/ sternum
C. thorax/ coccyx
D. sacrum / sternum
- The appendicular skeleton _____ the skeleton of the upper and lower limbs.
A. makes up
B. consists
C. forms
D. A and C
- The arm bone is called the _____; the forearm bones are the _____ and the radius.
A. fibula / ulna
B. humerus/ tibia
C. humerus / ulna
D. tibia/ fibula
- There are eight _____ in each wrist, five _____ in each palm, three phalanges in each finger, and two phalanges in each thumb.
A. carpals/ metacarpals
B. carpals / metatarsals
C. tarsals/ metatarsals
D. tarsals/ metacarpals
- There are seven _____ in the ankle, five _____ in the middle of each foot, two phalanges in each big toe, and three phalanges in each other toe.
A. carpals/ metacarpals
B. carpals / metatarsals
C. tarsals/ metatarsals
D. tarsals/ metacarpals
- _____ their rigid nature, bones provide a framework for the body.
A. Because of
B. Due to
C. Owing to
D. All are correct.
- Together with skeletal muscles, tendons, ligaments and joints, the bones form the _____ machinery of human body.
A. move
B. moving
C. moved
D. moves
- All long bones have the red bone marrow, _____ hematopoietic stem cells, capable of producing blood cells.
A. that contains
B. which contain
C. which contains
D. that contain

Multiple Choice Quiz Skeletal Muscular System

JR Anderson



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web k sp of calcium hydroxide 1 author j m mccormick last update ocober 13 2013 introduction an equilibrium constant k is related to ΔG for a process through eqn 1 and as long as ΔH and ΔS are independent of temperature eqn 1 can be combined with eqn 2 to determine ΔH and ΔS from the temperature dependence of k

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web your primary objective in this experiment is to test a saturated solution of calcium hydroxide and use your observations and measurements to calculate the ksp of the compound you will do this by titrating the prepared ca oh 2 solution with a standard hydrochloric acid solution

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of ca oh is represented in equation form as shown below ca oh s ca aq 2oh aq the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is ca oh 2 ca2 2 oh

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oh 2 m ca in saturated ca oh 2 m ksp of ca oh 2 average ksp of ca oh 2

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web the chemical equation used was $\text{Ca(OH)}_2 \text{ (s)} \rightleftharpoons \text{Ca}^{2+} \text{ (aq)} + 2\text{OH}^{-} \text{ (aq)}$ the solubility product was determined to be $K_{sp} \text{ Ca(OH)}_2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3\text{PO}_4\text{ (s)}$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore $K_{sp} \text{ Ca}_3\text{PO}_4\text{ (s)} = [\text{Ca}^{2+}]^3 [\text{PO}_4^{3-}]^2$

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web therefore it was determined through error calculations and ease of usethat the most reliable valid method to use to get the closest ksp value tothe theoretical ksp for ca was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the K_{sp} for calcium hydroxide $K_{sp} Ca^{2+} OH^{-} 2 \times 10^{-2}$ $K_{sp} 5 \times 10^{-4}$ find the accepted value of the K_{sp} for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error accepted value for K_{sp} for sodium hydroxide is 5×10^{-6}

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