

### 18.4 Acid-Base Chemistry: Theory

#### Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct) \_\_\_\_\_  
 (A) pressure, concentration, (B) pressure, pH  
 (C) molar fraction, (D) dimensional analysis
2. Which of the following is CORRECT? (dimensionless or they dimensional quantities?)

(A)  $\ln \frac{1}{10}$  (B)  $\ln 10$

(C)  $\ln \frac{1}{10} \times 2.303$  (D)  $\ln 10 \times 2.303$  (E)  $\ln \frac{1}{10} \times 2.303 \times 10$  (F) none of the above

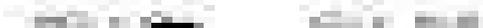
3. In a given solution, the following equilibrium constant and initial amounts, at 25°C, are assigned:  $K_a = 10^{-4}$  mol/L, and only 10% of the acid is ionized. What is the initial concentration of the acid?

(A)  $10^{-4}$  mol/L (B)  $10^{-3}$  mol/L (C)  $10^{-2}$  mol/L (D)  $10^{-1}$  mol/L (E)  $10^0$  mol/L

4. In which reaction does the  $\Delta H_{\text{rxn}}$  value for the reaction involve the largest amount of energy? (correct) \_\_\_\_\_



5. Which of the following is proper for the following equation? \_\_\_\_\_



(A)  $\Delta H_{\text{rxn}} < 0$  (B)  $\Delta H_{\text{rxn}} > 0$  (C)  $\Delta H_{\text{rxn}} = 0$  (D)  $\Delta H_{\text{rxn}} = 1$

6. A solution of acetic acid and its conjugate base is prepared with initial amounts as shown below. If a 10% of acetic acid is ionized and 10% of the conjugate base is ionized, what is the pH of the solution? \_\_\_\_\_



(A) 10.00 (B) 1.00 (C) 2.00 (D) 3.00 (E) 4.00

7. In a full equilibrium, acetic acid, concentration is 0.1 mol/L, and its conjugate base concentration is 0.1 mol/L. After the reaction was completed, the equilibrium constant was 10.0. What is the initial concentration of the acetic acid? \_\_\_\_\_



# Multiple Choice Stoichiometry Test

**Emilie Sanchez**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

**Multiple Choice Stoichiometry Test:**

The book delves into Multiple Choice Stoichiometry Test. Multiple Choice Stoichiometry Test is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Multiple Choice Stoichiometry Test, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Multiple Choice Stoichiometry Test
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- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Multiple Choice Stoichiometry Test. This chapter will explore what Multiple Choice Stoichiometry Test is, why Multiple Choice Stoichiometry Test is vital, and how to effectively learn about Multiple Choice Stoichiometry Test.

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6. In chapter 5, this book will draw a conclusion about Multiple Choice Stoichiometry Test. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Multiple Choice Stoichiometry Test.

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