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textbook if you get stuck try asking another group for help 1 for the reaction below if substance a is disappearing at a rate of  $1.82 \times 10^{-2} \text{ mol l}^{-1} \text{ s}^{-1}$  at what rate is c appearing  $3a + 3b \rightarrow 5c + 2d$

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web reaction rates google classroom  $mg + s + 2hcl \rightarrow aq + mgcl_2 + 2aq + h_2g$  the reaction between  $mg + s$  and  $hcl + aq$  is represented by the equation above in a kinetics experiment a  $0.080 \text{ g}$  sample of  $mg + s$  ribbon is added to an erlenmeyer flask containing  $50 \text{ ml}$  of  $3.0 \text{ M}$   $hcl + aq$  at  $25 \text{ }^\circ\text{C}$

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