

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities (correct)?
 (A) pressure, concentration (B) pressure, pH
 (C) molar fraction (D) dimensional analysis
2. Which of the following is CORRECT? (A) concentration of ions increases as pH increases (B) concentration of ions increases as pH decreases



3. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. If the pH of the solution is 4.0, what is the concentration of acetic acid?



4. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. What is the pH of the solution?



5. Which of the following is CORRECT? (A) pH of a 0.1 M acetic acid solution is 4.0 (B) pH of a 0.1 M acetic acid solution is 5.0 (C) pH of a 0.1 M acetic acid solution is 6.0 (D) pH of a 0.1 M acetic acid solution is 7.0



6. A solution of acetic acid and its conjugate base is prepared with a pH of 4.0. If the concentration of acetic acid is 0.1 M, what is the concentration of acetate ions?



7. In a 0.1 M acetic acid solution, the concentration of acetate ions is 0.01 M. What is the pH of the solution?



Multiple Choice Stoichiometry Test

Jicheng Xie



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