

17.5

HYDROGEN IONS AND ACIDITY

Section Overview

Objectives

1. Calculate concentrations based on the pH of acids and bases and vice versa.
2. Calculate pH values and concentrations of acids and bases and vice versa.
3. Calculate the pH of weak acids and bases.

Vocabulary

- | | | |
|-------------------|------------------------------|--------------------|
| 1. autoionization | 2. conjugate acid of a base | 3. pH |
| 4. autoionization | 5. conjugate base of an acid | 6. polyprotic acid |
| | 7. weak acid | |

Key Equations

- | | |
|------------------------------------|----------------------------------------------------------------------------------|
| 1. $\text{pH} = -\log[\text{H}^+]$ | 2. $\text{pH} = \text{pK}_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$ |
| 3. $\text{pH} = -\log[\text{H}^+]$ | 4. $\text{pH} = \text{pK}_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$ |

Part A: Completion

Use the appropriate number to identify the relationship of the conjugate acid-base pair indicated in the second step of the acid-base reaction shown in the following equation.

1. _____ is the conjugate base of _____.
2. _____ is the conjugate acid of _____.
3. _____ is the conjugate base of _____.
4. _____ is the conjugate acid of _____.
5. _____ is the conjugate base of _____.
6. _____ is the conjugate acid of _____.
7. _____ is the conjugate base of _____.
8. _____ is the conjugate acid of _____.
9. _____ is the conjugate base of _____.
10. _____ is the conjugate acid of _____.

- | | |
|-----|-------|
| 1. | _____ |
| 2. | _____ |
| 3. | _____ |
| 4. | _____ |
| 5. | _____ |
| 6. | _____ |
| 7. | _____ |
| 8. | _____ |
| 9. | _____ |
| 10. | _____ |

19 2 Hydrogen Ions And Acidity Worksheet Answers

R Sandford



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