

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) volume of precipitate, concentration of ions, concentration of precipitate

(A) 10^{-14} (B) 10^{-7}

(A) 10^{-14} (B) 10^{-7} (C) 10^{-14} (D) 10^{-7} (E) 10^{-14} (F) 10^{-7} (G) 10^{-14} (H) 10^{-7} (I) 10^{-14} (J) 10^{-7}

2. In a 0.1 M solution of a weak acid, the concentration of the acid is 0.09 M. What is the pH of the solution? (correct)

(A) 1.0 (B) 1.1 (C) 1.2 (D) 1.3 (E) 1.4

(A) 1.0 (B) 1.1 (C) 1.2 (D) 1.3 (E) 1.4 (F) 1.5 (G) 1.6 (H) 1.7 (I) 1.8 (J) 1.9

3. The pH of a 0.1 M solution of a weak acid is 1.0. What is the concentration of the acid? (correct)

(A) 0.1 (B) 0.01 (C) 0.001 (D) 0.0001 (E) 0.00001

4. Which of the following is a weak acid? (correct)



(A) 10^{-14} (B) 10^{-7} (C) 10^{-14} (D) 10^{-7} (E) 10^{-14} (F) 10^{-7} (G) 10^{-14} (H) 10^{-7} (I) 10^{-14} (J) 10^{-7}

5. A 0.1 M solution of a weak acid has a pH of 1.0. What is the concentration of the acid? (correct)

(A) 0.1 (B) 0.01 (C) 0.001 (D) 0.0001 (E) 0.00001

(A) 0.1 (B) 0.01 (C) 0.001 (D) 0.0001 (E) 0.00001 (F) 0.000001 (G) 0.0000001 (H) 0.00000001 (I) 0.000000001 (J) 0.0000000001

6. In a 0.1 M solution of a weak acid, the concentration of the acid is 0.09 M. What is the pH of the solution? (correct)

(A) 1.0 (B) 1.1 (C) 1.2 (D) 1.3 (E) 1.4 (F) 1.5 (G) 1.6 (H) 1.7 (I) 1.8 (J) 1.9

Multiple Choice Stoichiometry Test

Xiaolong Qi



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