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they are the same as basic solutions

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nh<sub>2</sub> and s<sup>2-</sup> are ch<sub>4</sub> nh<sub>3</sub> and hs respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that s<sup>2-</sup> is the weakest base

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compounds have characteristics of both acids and bases in the same molecule we will explore the relationship between

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