

### Lesson Plan

#### Chapter 2 : The Sociological Perspective of the self

Prepared by: Noella O. Ursal

#### I. Learning objectives

At the end of the lesson, the students are expected to:

1. Recognize what sociology tells about understanding the self and others.
2. Explain the theories of the social self.
3. Reflect on the importance of our society in developing our self.

#### II. Subject matter

A. **Topic** : The Sociological Perspective of the self

B. **References** : Simon, J [JehnSimon] 2022. *UTS (Understanding the Self) Lesson 2: Sociological Perspective of the Self*. <http://youtu.be/vzXFAiXnfGU?>

C. **Materials** : PowerPoint Presentation

D. **Values Integrated** : Cooperation

#### III. Procedure

	Teacher's Activity	Student's Activity
A. Preliminary Activities -Greetings	Good Morning Class	Good Morning Ma'am
Opening Prayer		
Classroom management		
Checking of Attendance	Let us first check your attendance for today. Is there any absent today?	There's no absent ma'am.
Introduction / Motivation		
Instruction / Delivery	<p><b>Sociology</b> as a science, study social groups and human relationships which help shed new insight into the interconnectedness between the self and other people.</p> <p><b>Charles Horton Cooley</b>, the self is not dependent on biological predispositions; rather, it is a product of social interaction.</p> <p>One of the pioneering contributions to sociological perspectives was Charles Cooley (1864-1929). He asserted that people's self-understanding is constructed, in part, by their perception of how others view them – a process term "the looking glass self".</p> <p>Sociologist <b>Charles Horton Cooley</b> introduced <b>the looking-glass self</b> to highlight that the people whom a person interacts become a mirror in which he views himself.</p> <p><b>The looking Glass Self (Cooley, 1902)</b></p>	Students will give their insights about sociological perspective of self.

# Sociological Perspective Lesson Plan

**Wolfgang Guggemos**



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta H = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta H = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) \rightarrow \frac{7}{2} \text{O}_2(\text{g}) + 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta H = 283 \text{ kJ}$

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web thus for reaction 9 2 in which 1 mole of  $h_3o$  and 1 mole of  $oh$  combine to form 2 moles of  $h_2o$   $\Delta H = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta H_{products} - \Delta H_{reactants}$  is positive if heat is absorbed that is if  $\Delta H_{products}$

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